



Mitochondria generated nitric oxide protects against permeability transition via formation of membrane protein S-nitrosothiols

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ABSTRACT

Mitochondria generated nitric oxide (NO^{*}) regulates several cell functions including energy metabolism, cell cycling, and cell death. Here we report that the NO^{*} synthase inhibitors (L-NAME, L-NNA and L-NMMA) administered either *in vitro* or *in vivo* induce Ca²⁺-dependent mitochondrial permeability transition (MPT) in rat liver mitochondria via a mechanism independent on changes in the energy state of the organelle. MPT was determined by the occurrence of cyclosporin A sensitive mitochondrial membrane potential disruption followed by mitochondrial swelling and Ca²⁺ release. In *in vitro* experiments, the effect of NOS inhibitors was dose-dependent (1 to 50 μM). In addition to cyclosporin A, L-NAME-induced MPT was sensitive to Mg²⁺ plus ATP, EGTA, and to a lower degree, to catalase and dithiothreitol. In contrast to L-NAME, its isomer D-NAME did not induce MPT. L-NAME-induced MPT was associated with a significant decrease in both the rate of NO^{*} generation and the content of mitochondrial S-nitrosothiol. Acute and chronic *in vivo* treatment with L-NAME also promoted MPT and decreased the content of mitochondrial S-nitrosothiol. SNAP (a NO^{*} donor) prevented L-NAME mediated MPT and reversed the decrease in the rate of NO^{*} generation and in the content of S-nitrosothiol. We propose that S-nitrosylation of critical membrane protein thiols by NO^{*} protects against MPT.

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1. Introduction

Nitric oxide (NO^{*}) is an ubiquitous intra and inter-cellular messenger involved in the regulation of diverse physiological processes, such as neurotransmission, vasodilation, immune response and cell cycling and arrest [1,2]. On the other hand, at elevated concentrations, NO^{*} can also be cytotoxic [3,4]. The synthesis of NO^{*} from L-arginine and O₂ is catalyzed by nitric oxide synthases (NOS). Up to the early 90s, three distinct isoforms of NOS had been well characterized, two constitutive (endothelial, eNOS, and neuronal, nNOS) and one inducible (iNOS). They require several co-factors, including

FAD, FMN, BH₄, Ca²⁺ and NADPH. All NOS isoforms require Ca²⁺ for their activity, although iNOS is significantly less dependent on Ca²⁺ than the other isoforms [5].

The existence of a mitochondrial NO^{*} synthase (mtNOS) was first suggested by immunohistochemical studies that found that anti-NOS antibodies stained mitochondria [6–8]. Next, biochemical properties of mtNOS were characterized in purified liver mitochondria [9–11]. The activity of mtNOS was also determined in mitochondria purified from different tissues of rats and mice, including brain, heart, kidney, thymus and skeletal muscle (for a review see [12]). The identity of a liver mtNOS was described by Elfering et al. [13]. They sequenced mtNOS, identified the protein as a splice variant of the nNOS, and localized it in the inner mitochondrial membrane. In support for this finding, a previous work described the lack of mtNOS in the heart of nNOS knockout mice [14]. More recent evidences suggest that mtNOS is structurally linked to both complexes I and IV [15,16]. Nonetheless, a considerable debate about the molecular identity of the mtNOS persists [17–20]. Additionally, NO^{*} can be produced inside mitochondria by other NOS independent pathways [21–23].

The mitochondria generated NO^{*} regulates O₂ consumption, energy metabolism, and reactive oxygen species (ROS) formation (for a review see [24]). Mitochondria also generate the anion superoxide (O₂^{•-}). Although these two radical species are not toxic at physiological concentrations, an imbalance in their production or

Abbreviations: Alam, alamethicin; CsA, cyclosporin A; DAF-FM, (4-amino-5-methylamino-2',7'-difluorofluorescein) diacetate; D-NAME, N^G-Nitro-D-arginine methyl ester hydrochloride; DTT, dithiothreitol; EGTA, ethylene glycol-bis(β-aminoethyl ether)-N,N,N',N'-tetraacetic acid tetrasodium salt; GSH, glutathione; L-NAME, N^G-nitro-L-arginine methyl ester hydrochloride; L-NMMA, N^G-methyl-L-arginine acetate salt; L-NNA, N⁵-(Nitroamidino)-L-2,5-diaminopentanoic acid; MPT, mitochondrial permeability transition; NO^{*}, nitric oxide; NOS, nitric oxide synthase; PTP, permeability transition pore; RLM, rat liver mitochondria; RNS, reactive nitrogen species; ROS, reactive oxygen species; SNAP, S-nitroso-N-acetyl-DL-penicillamine; SOD, superoxide dismutase

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elimination can be responsible for alterations of molecular mechanisms regulating cell life. The reaction of these two species generates peroxynitrite (ONOO^-), a potent oxidant that reacts with most biological molecules causing cell damage. In addition to peroxynitrite, other reactive nitrogen species [6] are derived from NO^* , including nitroxyl (HNO), nitrosonium cation (NO^+) and S-nitrosothiols (RSNOs), [25–29].

NO^* reversibly interacts with cytochrome oxidase modulating mitochondria respiration rate [2,30]. In addition, the interaction of NO^* with thiol groups forming S-nitrosothiols in mitochondria may be of physiological importance [31]. The formation of protein S-nitrosothiols may enable the reversible regulation of protein function [32]. The biological role of S-nitrosylation is also related to the protection of protein thiols under oxidative stress conditions, as it will be discussed ahead.

Mitochondrial protein oxidation, lipid peroxidation and DNA damage caused by ROS and RNS have been extensively documented (reviewed in [28,33,34]). The inner membrane proteins are primary targets for oxidative damage, a process that is activated by excessive matrix Ca^{2+} . Oxidation of membrane protein thiols induces the formation of the permeability transition pore (PTP) via protein aggregation due to dithiol formation [35]. We have proposed that these protein aggregates form aqueous channels that confer non-selective permeabilization of the inner mitochondrial membrane [36]. A large body of evidences indicates that mitochondrial permeability transition is involved in cell death mechanisms determining whether the fate of cells will be apoptosis or necrosis [37]. It seems to operate as an ATP switch by establishing the balance between ATP availability and ATP hydrolysis after PTP opening [38].

It is well known that NO^* has pro- and anti-apoptotic effects depending on its concentration range. At high production rates ($>2 \mu\text{M/s}$), NO^* induces MPT via a Ca^{2+} -dependent mechanism, probably due to the formation of ONOO^- . This RNS reacts with thiols by one or two electron oxidations to produce thiol radicals (RS^*) and sulfenic acids (RSOH), respectively, both of which usually react with protein thiols to form disulfides or other forms of higher thiol oxidation [31]. In contrast, at low levels, NO^* may inhibit MPT (IC_{50} of $11 \text{ nM NO}^*/\text{s}$) [39].

The objective of the present study was to investigate the relationship between mitochondria generated NO^* and MPT using non-selective NOS inhibitors under experimental conditions of fully preserved mitochondrial respiration. We tested the hypothesis that NO^* protects against MPT via S-nitrosylation of critical mitochondrial membrane protein thiol groups.

2. Materials and methods

2.1. Isolation of liver mitochondria

Liver mitochondria from adult rats fasted overnight were isolated by conventional differential centrifugation [40] and partially purified by a discontinuous Percoll gradient. Briefly, livers were homogenized in 250 mM sucrose, 1 mM EGTA, and 10 mM Hepes buffer (pH 7.2), and centrifuged for 10 min at 2500 rpm. The supernatant was centrifuged for 10 min at 8000 rpm. The mitochondrial pellet was purified using a discontinuous Percoll gradient according to [41]. The mitochondrial fraction obtained from the interface between 19 and 52% Percoll layers was washed in medium containing 250 mM sucrose, 10 mM Hepes and 0.1 mM EGTA, for 10 min at 8000 rpm. The final pellet was resuspended in 250 mM sucrose and 10 mM Hepes to a final protein concentration of 80–100 mg/ml.

2.2. Standard incubation procedure

The experiments were carried out at 28°C , with continuous magnetic stirring, in standard medium containing 125 mM sucrose,

65 mM KCl, 2 mM inorganic phosphate, 1 mM magnesium chloride, 10 mM Hepes buffer (pH 7.2) and a complex I substrate mixture made of malate, pyruvate, ketoglutarate and glutamate (5 mM each). Other additions are indicated in the figure legends.

2.3. Animals and L-NAME treatment

Female Wistar rats (180–250 g) had free access to standard laboratory rodent chow diet (Nuvital CR1, PR, Brazil) and water and were housed at $22 \pm 2^\circ\text{C}$ on a 12 h light–dark cycle. L-NAME (50 mg/kg i.p.) or saline was given once i.p., and after 14 h, rats were killed by cervical dislocation, livers were perfused with cold saline solution and excised for mitochondria isolation. Some rats were also subject to a chronic treatment with L-NAME added to their drinking-water ($1 \text{ mg kg}^{-1} \text{ day}^{-1}$) during 14 days.

2.4. Mitochondrial swelling

Mitochondrial swelling (0.5 mg/ml) was determined as the decrease in the turbidity of the mitochondrial suspension measured at 520 nm using a temperature-controlled Hitachi U-3000 spectrophotometer (28°C).

2.5. Mitochondrial membrane potential disruption

Mitochondria (0.5 mg/ml) were incubated in the standard reaction medium containing $2 \mu\text{M}$ tetraphenylphosphonium (TPP^+) at 28°C . The concentration of TPP^+ in the extramitochondrial medium was continuously monitored with a TPP^+ -selective electrode prepared in our laboratory according to [42]. The membrane potential was then calculated by assuming that the TPP^+ distribution between mitochondria and the medium follows the Nernst equation. Corrections due to the binding of TPP^+ to the mitochondrial membrane were made according to Jensen et al. [43]. No corrections were made to compensate $\Delta\psi$ values for the continuous alteration in mitochondrial volume that occurs during the experiments.

2.6. Mitochondrial calcium transport

Calcium uptake and release by isolated liver mitochondria (0.5 mg/ml) was determined following the fluorescence of $0.1 \mu\text{M}$ Calcium Green-5N hexapotassium salt (Molecular Probes) using a temperature-controlled Hitachi F4500 spectrofluorometer (28°C) at excitation and emission wavelengths of 506 and 531 nm, respectively, and slit widths of 5.0 nm [44].

2.7. Mitochondrial nitric oxide production

The probe DAF-FM ($10 \mu\text{M}$) was used to monitor NO^* release from rat liver mitochondria (0.5 mg/ml) using a temperature-controlled Hitachi F4500 spectrofluorometer (28°C) at excitation and emission wavelengths of 495 and 515 nm, respectively, and slit widths of 2.5 nm. Superoxide dismutase ($1 \mu\text{M}$) and catalase ($1 \mu\text{M}$) were added to prevent interference of $\text{O}_2^{\bullet-}$ and H_2O_2 . To test DAF specificity to NO^* under our experimental conditions we added the NO^* donor SNAP at different concentrations, and observed a rapid and proportional increase in the probe fluorescence. The calibration was made using DAF-2T the product of the reaction between DAF-FM and NO^* .

2.8. Mitochondrial S-nitrosothiol determination

The samples were prepared according to [3]. Mitochondria were disrupted by shaking with 0.5% Triton X-100 for 60 min at 4°C and insoluble proteins were removed by centrifugation at 20,000 g for 60 min. The supernatant was used to determine S-nitrosothiol

content using a modified Saville assay [45]. Briefly, the samples were incubated with 300 μM 2,3-diaminonaphthalene (DAN) diluted in HCl (0.5 M). HgCl_2 was used to release NO^* during a 10 min incubation in the dark. After this, samples were alkalized with NaOH (1 M) and incubated for an additional 10 min period in the dark. Then, fluorescence was followed in a temperature-controlled Hitachi F4500 spectrofluorometer (28 °C) at excitation and emission wavelengths of 381 and 407 nm, respectively, and slit widths of 2.5 nm (excitation) and 5.0 nm (emission). Nitrosothiol content was calculated using a standard curve generated with NaNO_2 .

2.9. Statistical analyses

Data from the experiments were analyzed by one-way ANOVA, followed by a post hoc Tukey test, using OriginLab 7.5 software (OriginLab Corp.). Student *t*-test was used for two mean comparisons. The data are presented as averages \pm SD of at least four experiments conducted with different preparations.

2.10. Chemicals

L-NAME, D-NAME, L-NNA, L-NMMA, SNAP, Cyclosporin A, catalase, SOD, EGTA, FCCP, NEM, HgCl_2 , 2,3-Diaminonaphthalene (DAN), and NaNO_2 were purchased from Sigma (St. Louis, USA). Calcium green-5N and DAF-FM were purchased from Molecular Probes. DAF-2T was obtained from Alexis Biochemicals. All other reagents were products of the highest purity grade available.

3. Results

The effects of NOS inhibitors on mitochondrial swelling and membrane potential are shown in Fig. 1. The experiments depicted in Fig. 1A show that L-NAME, L-NNA and L-NMMA (50 μM each), in contrast to the L-NAME isomer D-NAME, induced extensive swelling of mitochondria suspended in the incubation medium containing 30 μM Ca^{2+} . The effect of L-NAME was dose-dependent (1–50 μM) both on mitochondrial swelling (Fig. 1B) and on the disruption of membrane potential (Fig. 1C).

The mechanisms by which L-NAME induced the loss of mitochondrial integrity *in vitro* were additionally investigated. Fig. 2 shows that the effects of L-NAME on mitochondrial swelling (panel A) and membrane potential disruption (panel B) were inhibited, totally or partially, by the MPT inhibitors CsA, Mg^{2+} plus ATP, catalase, and DTT, a disulfide reductant.

Since MPT is stimulated by Ca^{2+} accumulation and the inhibitory effects of NO^* on MPT could be secondary to inhibition of Ca^{2+} uptake by the organelle [39], we examined the effect of L-NAME on mitochondrial Ca^{2+} handling, under our experimental conditions (Fig. 2C). The data show that initially the mitochondrial Ca^{2+} influx was unaffected by the presence of L-NAME. However, after a Ca^{2+} pulse, the capacity to retain Ca^{2+} was decreased by L-NAME, via a mechanism sensitive to cyclosporin A. The results indicate that L-NAME, as well as the other NOS inhibitors, the arginine analogues L-NNA and L-NMMA, has the ability to act as a MPT inducer independently of changes in the capacity of mitochondrial Ca^{2+} accumulation and respiration (data not shown).

To investigate whether a NO^* donor could prevent the effect of L-NAME on MPT, we performed experiments in the presence of SNAP (Fig. 3). Interestingly, Fig. 3A shows that SNAP, at the concentration of 10 nM, retards both the swelling induced by Ca^{2+} alone and that induced by Ca^{2+} plus L-NAME. The presence of 10 nM SNAP did not affect Ca^{2+} influx to mitochondria but, in agreement with the experiment of Fig. 3A, retards Ca^{2+} release in the presence or absence of L-NAME (Fig. 3B). These results suggest that neither L-NAME nor 10 nM SNAP have a direct effect on mitochondrial permeabilization and that the observed effects are cer-

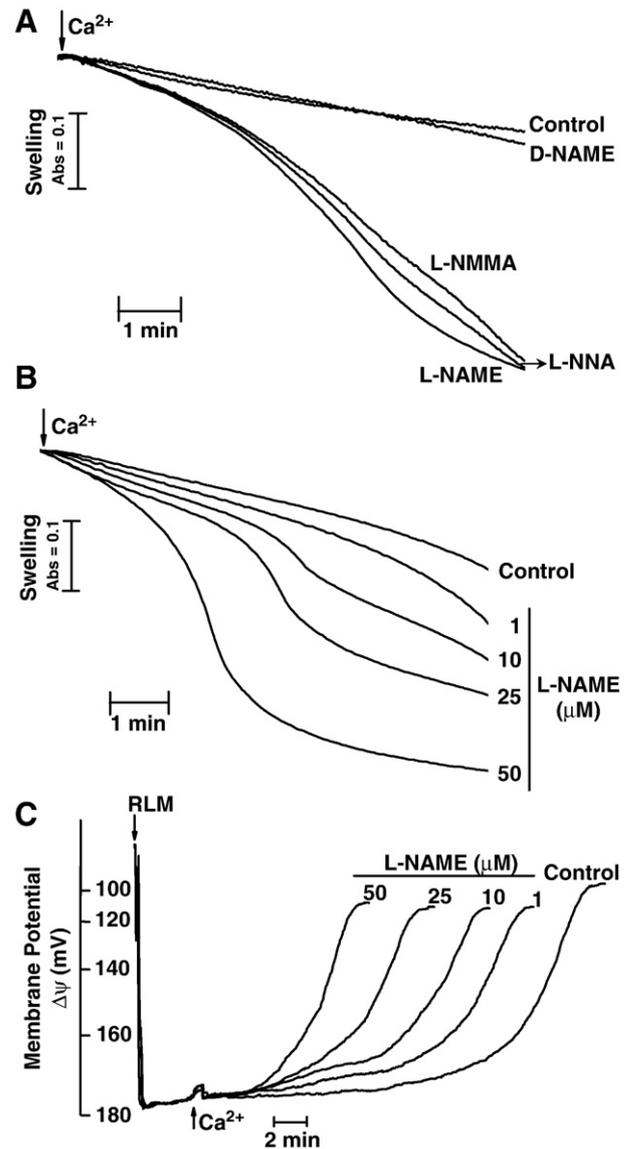


Fig. 1. Mitochondrial swelling induced by NOS inhibitors *in vitro* (A) and dose-dependent effect of L-NAME on mitochondrial swelling (B) and on mitochondrial membrane potential ($\Delta\psi$) disruption (C). Panel A: Rat liver mitochondria (RLM, 0.5 mg/ml) were incubated in the standard medium containing 30 μM Ca^{2+} in the presence of 50 μM of L-NAME, L-NMMA, L-NNA, or D-NAME. Absorbances at 7 min were: 1.270 ± 0.062 (Control), 1.253 ± 0.093 (D-NAME), 0.955 ± 0.075 (L-NAME), 0.958 ± 0.037 (L-NNA), and 0.966 ± 0.186 (L-NMMA), $p < 0.001$ for control and D-NAME vs. all NOS inhibitors. Panel B: L-NAME was present at indicated concentrations. Absorbances at 7 min were: 1.465 ± 0.104 (Control), 1.368 ± 0.196 (1 μM L-NAME), 1.175 ± 0.172 (10 μM L-NAME), 1.145 ± 0.167 (25 μM L-NAME), and 0.948 ± 0.023 (50 μM L-NAME), $p < 0.05$ for control vs. 10, 25 and 50 μM L-NAME. Panel C: Standard medium containing 2 μM TPP^+ , 25 μM Ca^{2+} and L-NAME at indicated concentrations. Average time of $\Delta\psi$ maintenance was: 24.9 ± 2.8 (Control), 18.6 ± 1.5 (1 μM L-NAME), 16.3 ± 7.3 (10 μM L-NAME), 14.8 ± 3.0 (25 μM L-NAME) and 12.7 ± 1.9 (50 μM L-NAME), $p < 0.05$ for control vs. 10, 25 and 50 μM L-NAME. Lines are representative of five independent experiments.

tainly mediated by NO^* , in a concentration range that does not affect either Ca^{2+} influx or respiration (not shown).

In order to further characterize the role of endogenous NO^* on MPT, we performed experiments analyzing the effect of L-NAME on NO^* generation by mitochondria (Fig. 4). The experiments started by the addition of mitochondria to the standard medium containing 10 μM of the NO^* probe, DAF-AM (Fig. 4A). A biphasic increase in DAF fluorescence was observed: an initial phase with higher slope was followed by a second phase with lower slope that began 3–4 min after

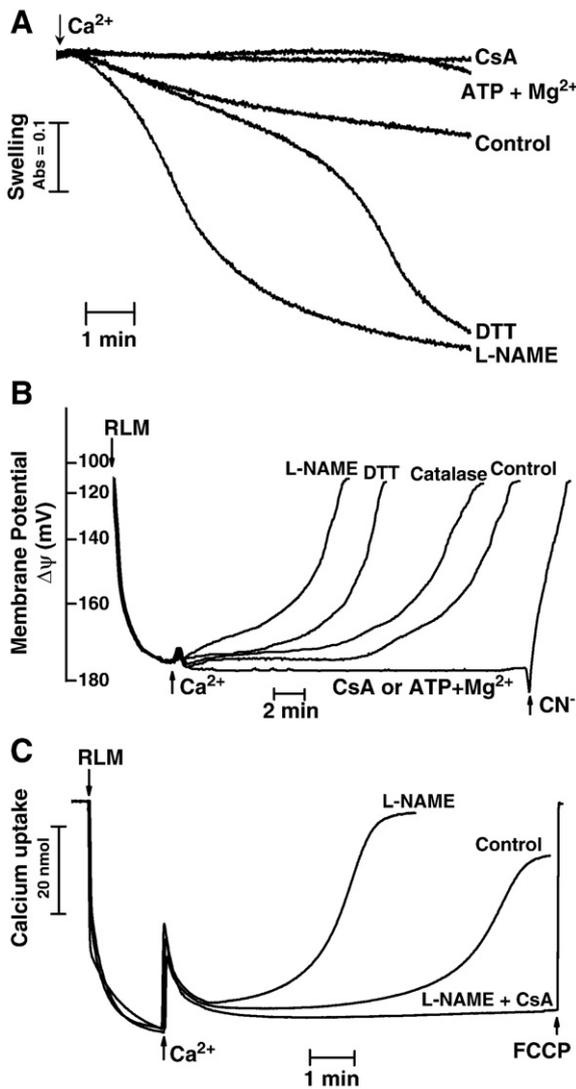


Fig. 2. Inhibition of L-NAME-induced mitochondrial swelling (A), membrane potential disruption (B) and calcium release (C) by MPT inhibitors. Rat liver mitochondria (RLM, 0.5 mg/ml) were incubated in the standard medium with 50 μ M L-NAME and 30 μ M Ca^{2+} in the presence of the MPT inhibitors: cyclosporin A (CsA, 1 μ M), ATP (200 μ M) plus Mg^{2+} (5 mM), catalase (2 μ M), or dithiothreitol (DTT, 500 μ M). Panel A: Absorbances at 7 min were: 1.332 ± 0.138 (Control), 1.395 ± 0.172 (CsA), 1.387 ± 0.137 (ATP + Mg^{2+}), 1.020 ± 0.047 (DTT), and 0.980 ± 0.043 (L-NAME), $p < 0.05$ for L-NAME vs. control, CsA, and ATP + Mg^{2+} . Panel B: Average time of $\Delta\psi$ retention (min) was: 24.0 ± 5.4 (Control), 18.8 ± 3.5 (DTT), 20.8 ± 1.8 (Catalase) and 14.9 ± 3.6 (L-NAME), $p < 0.05$ for L-NAME vs. control, CsA, and ATP + Mg^{2+} . Panel C: Average time of calcium release (min) was: 7.1 ± 0.6 (Control) and 2.8 ± 1.0 (L-NAME), $p < 0.05$. Lines are representative of five independent experiments.

mitochondria addition. The first faster phase of fluorescence increase may reflect the reaction of DAF with a pre-formed mitochondrial pool of NO^* and the second slower phase of fluorescence increase probably represents the steady state rate of NO^* generation by mitochondria. This second phase was considered for the estimations of mitochondrial NO^* generation rates. According to our standard curve built using DAF-2T, the rates of NO^* generation were 0.35 pmol $NO^*/$ mg under the control conditions and 0.18 pmol $NO^*/$ mg in the presence of L-NAME (Fig. 4B). Therefore, 50 μ M L-NAME caused an inhibition of 47% in the mitochondria generation rate of NO^* . Furthermore, an experiment using D-NAME, instead of L-NAME, did not show any difference in the rate of DAF fluorescence as compared to the control (Fig. 4B). Additions of increasing concentrations of the NO^* donor SNAP were done to the preparations containing L-NAME and

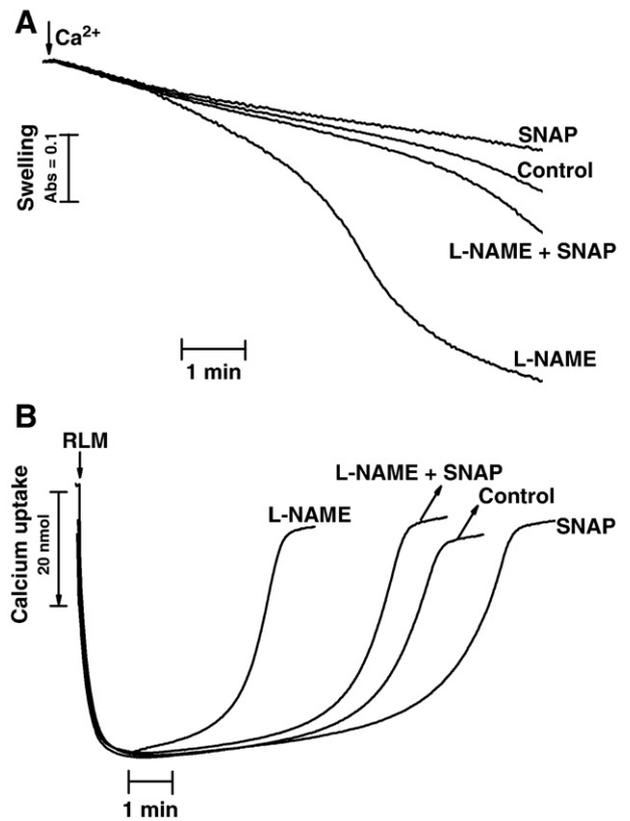


Fig. 3. SNAP protects against L-NAME-induced mitochondrial swelling (A) and calcium release (B). Rat liver mitochondria (RLM, 0.5 mg/ml) were incubated in standard medium containing 30 μ M Ca^{2+} in the presence or absence of SNAP (10 nM) and L-NAME (50 μ M) as indicated. Panel A: Absorbances at 7 min were: 1.327 ± 0.066 (SNAP), 1.295 ± 0.118 (Control), 1.250 ± 0.144 (L-NAME + SNAP) and 1.040 ± 0.121 (L-NAME), $p < 0.05$ for L-NAME vs. all others. Panel B: Average calcium retention time (min) was: 8.7 ± 0.9 (SNAP), 7.2 ± 0.5 (Control), 6.1 ± 2.0 (L-NAME + SNAP), and 2.9 ± 0.9 (L-NAME), $p < 0.05$ for L-NAME vs. all others. Lines are representative of five independent experiments.

significantly increased the slope of DAF fluorescence (Fig. 4A and B). No mitochondrial swelling was detected under these experimental conditions (low Ca^{2+} concentration) (data not shown).

Since NO^* promotes S-nitrosylation of cysteine residues in mitochondrial membrane proteins [4] we analyzed the influence of L-NAME on the content of mitochondrial S-nitrosothiol. Fig. 5 shows that L-NAME decreases S-nitrosothiol content by 36% when compared to the control. On the other hand, 10 nM SNAP totally abrogates the effect of L-NAME. In order to eliminate the possible interference of mitochondrial swelling in S-nitrosothiol estimation, we performed a control experiment in the presence of alamethicin, a non-selective inducer of mitochondrial membrane permeabilization. This compound did not cause any change in S-nitrosothiol content when compared to the control (Fig. 5), despite the extensive swelling it induces. The higher content of S-nitrosothiol found in our determinations in relation to literature data [30,46,47] is probably due to the fact that our assay was done in whole mitochondria, which presents a high content of matrix S-nitrosylated glutathione content [48].

The effect of L-NAME was also tested *in vivo*, using two treatment protocols that inhibit all isoforms of NOS in animal tissues. The experiments presented in Fig. 6A show that liver mitochondria isolated from rats 14 h after an acute dose of L-NAME (50 mg/kg i.p.) are more susceptible to MPT than those isolated from control animals. Under these experimental conditions, mitochondrial swelling was also prevented by the MPT inhibitors, CsA, ATP plus Mg^{2+} or catalase (Fig. 6B). More important, *in vivo* L-NAME treatment decreased mitochondrial

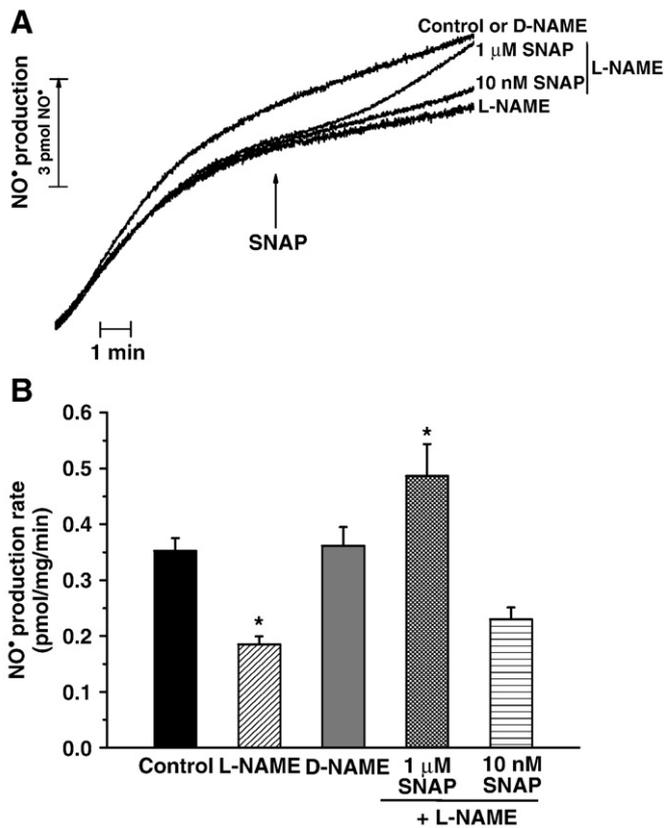


Fig. 4. Inhibition of mitochondrial NO* generation by L-NAME. NO* production by mitochondria (0.5 mg/ml) was measured using 10 μM DAF-FM in the presence of 1 μM SOD and 1 μM catalase. Standard medium contained 10 μM Ca²⁺. Panel A: Representative experiment of mitochondrial NO* production in the presence of D-NAME (50 μM) or L-NAME (50 μM), with addition of SNAP (10 nM or 1 μM) as indicated. Panel B: Quantification of total NO* generated (pmol/mg mitochondrial protein/min). **p*<0.05 for L-NAME and SNAP (1 μM) vs. control. Results represent six independent experiments.

S-nitrosothiol content by 45% when compared to the control (Fig. 7). Similar results were obtained when the rats were chronically treated with L-NAME added to their drinking-water (1 mg kg⁻¹ day⁻¹ during 14 days; results not shown).

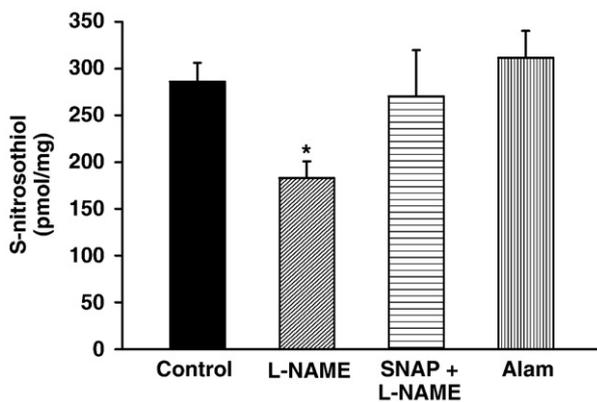


Fig. 5. SNAP protects against L-NAME-induced decrease in the content of mitochondrial S-nitrosothiol. Assay was performed in mitochondria (0.5 mg/ml) samples after a 10 min incubation in the presence of 30 μM Ca²⁺ and 50 μM L-NAME, 10 nM SNAP or L-NAME plus SNAP as indicated. Alamethicin (Alam, 5 μg/ml) plus EGTA (200 μM), a calcium independent mitochondria swelling inducer, was used to show no interference of mitochondrial swelling on the nitrosothiol quantification. **p*<0.05 for L-NAME vs. all others. Data are averages ± SD (*n*=8). Lines are representative of eight independent experiments.

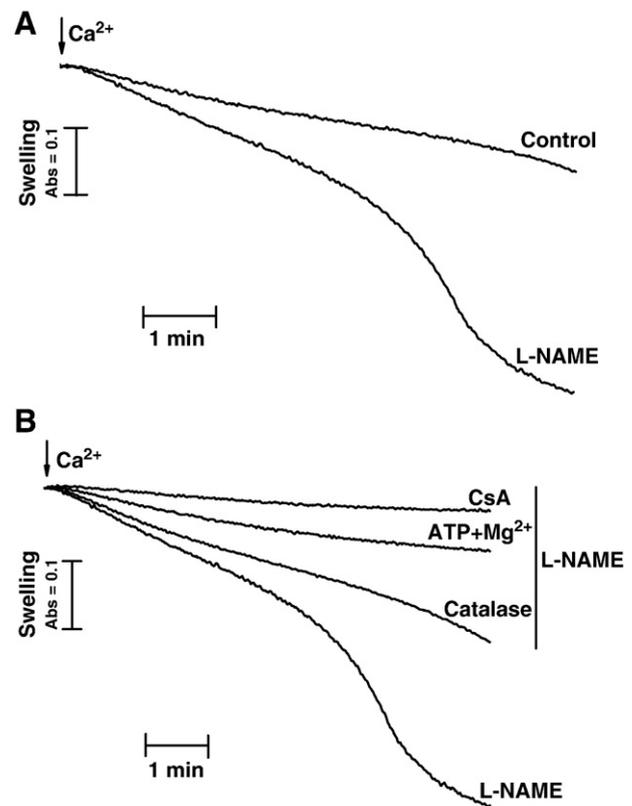


Fig. 6. *In vivo* treatment with L-NAME promotes Ca²⁺ induced mitochondrial swelling (A) sensitive to MPT inhibitors (B). Rat liver mitochondria (RLM, 0.5 mg/ml) from control and L-NAME treated rats were incubated in the presence of 30 μM Ca²⁺. Panel A: Absorbances for swelling at 7 min were: 1.300 ± 0.105 (Control) vs. 1.042 ± 0.187 (L-NAME), *p*<0.05. Panel B: RLM from L-NAME treated rats were incubated in the presence of MPT inhibitors cyclosporin A (CsA, 1 μM), ATP (200 μM) + Mg²⁺ (5 mM) and catalase (2 μM). Absorbances at 7 min were: 1.396 ± 0.029 (CsA), 1.324 ± 0.024 (ATP + Mg²⁺), 1.156 ± 0.104 (Catalase), and 1.062 ± 0.173 (L-NAME), *p*<0.05 for L-NAME vs. MPT inhibitors. Lines are representative of five independent experiments.

4. Discussion

The first evidences that mitochondria generate NO* and possess their own mtNOS were published more than 10 years ago [9–11]. Due to the enormous physiological relevance of NO*, the issue attracted great interest among researchers in the area and generated a large controversy. This subject has been critically analyzed in several recent

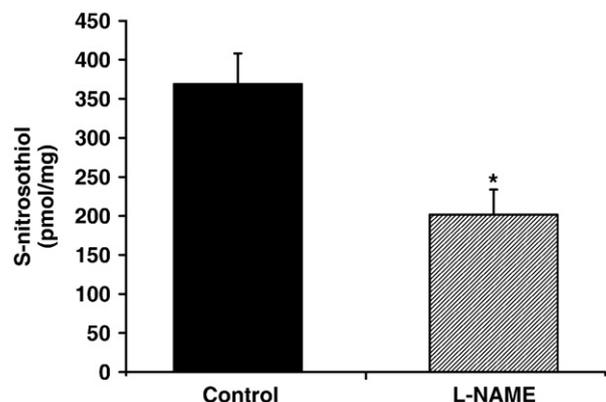


Fig. 7. *In vivo* treatment with L-NAME decreased the content of mitochondrial S-nitrosothiol. Assay was performed in mitochondria (0.5 mg/ml) samples after a 10 min incubation in the presence of 30 μM Ca²⁺, data are averages ± SD (*n*=5). **p*<0.05.

reviews [19,29,34,49]. Although the results reported in this work are not a final proof, they support the existence of such a mtNOS since they provide evidence that the lack of mitochondrially generated NO[•] impairs the mitochondrial resistance to Ca²⁺-dependent MPT. The previous demonstrations that Ca²⁺ induced MPT is sensitive to catalase and thiol reagents (mainly DTT) indicated that MPT is favored by accumulation of H₂O₂ and triggered by oxidation of membrane protein thiols [35,36,50,51]. The stimulation of MPT by NOS inhibitors, *in vitro* and *in vivo*, and conversely the protection against MPT conferred by the NO[•] donor SNAP in fully functioning mitochondria, are strong evidences of a new relevant physiological role of mtNOS, as opposed to others' claims [19].

In this study, the role of NO[•] on the MPT susceptibility was approached by using low concentrations of either NOS inhibitors or the NO[•] donor SNAP, which effectively modified the rates of NO[•] generation in the organelle. All non-selective NOS inhibitors used here, named L-NAME, L-NNA and L-NMMA showed the ability to induce Ca²⁺-dependent MPT by a dose-dependent mechanism antagonized by the NO[•] donor SNAP, in a process independent on changes of mitochondrial respiration rates and Ca²⁺ uptake. The effect of SNAP and lack of effect by D-NAME support the idea that these NOS inhibitors had no direct effect on MPT *per se*, but rather acted via changes in NO[•] availability in the organelle. Indeed, 50 μM L-NAME inhibited the rate of NO[•] generation by about 50% and this was totally reversed by 1 μM SNAP. The ability of these NOS inhibitors to promote MPT was also documented in mitochondria isolated from perfused livers of rats treated with L-NAME, either acutely or chronically. In addition, the lack of effect by D-NAME confirms the concept that all types of NOS are enantiomer specific for their amino acid substrate [52].

It is well established that NO[•] can have dual effects on mitochondrial functions. At physiological concentrations, it regulates some pathways such as metabolic rate [30,53], mitochondrial biogenesis [54,55], resistance to MPT [31,38,56] and cell death [57]. Recent work on mitochondria-target NO[•] donors shows protection during ischemia/reperfusion injury in the heart [30]. In contrast, high concentrations of NO[•] generate RNS and lead to pore opening [58], modify protein function, mainly through nitration of tyrosine residues [59,60] and impair lipid function also via nitration [59].

Concerning the mechanism of protection against MPT it is worth to consider that: first, many biological effects of NO[•] are mediated through S-nitrosylation of protein cysteine thiol and that mitochondria have been identified as organelles with high activity of reversible S-nitrosylation reactions [46]; second, PTP opening is mediated by membrane protein polymerization via dithiol formation [35,36]. In this regard, we observed that NOS inhibitors, under the experimental conditions that induced MPT, significantly decreased both the rate of NO[•] generation and the content of mitochondrial S-nitrosothiol. In contrast, SNAP that protected against MPT, recovered the rate of NO[•] generation and the content of S-nitrosothiols. Therefore, it is conceivable to propose that S-nitrosylation protects against MPT by competing for critical protein thiols and blocking the formation of dithiols. This putative role for NO[•] and S-nitrosothiols in the protection against MPT is in agreement with data showing that protein S-nitrosothiols can be generated by pre-formed S-nitrosothiols that persist mainly when the mitochondrial glutathione pool is oxidized [31,47], a condition that favors MPT [36]. This suggests that the reversible S-nitrosylation in mitochondria is related to the protection of protein thiols against the attack of reactive oxygen when the antioxidant mitochondrial system, mainly represented by NADPH and GSH, is deficient, a new biological role of mitochondrially generated NO[•].

In summary, we have identified that NOS inhibitors induce MPT through the decrease of both mitochondrial generation of NO[•] and S-nitrosylation of membrane protein thiols. These changes were shown to be independent on changes in respiration rates, membrane potential and Ca²⁺ content in the organelle. It is proposed that

S-nitrosylation of critical membrane protein thiols protects against MPT and thus against mitochondrial damage.

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